# Small-Angle X-ray Scattering – Interpretation of the results: "What can be obtained from the scattering curve?"

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#### Outline

#### Recapitulation

#### Self-assembled systems

Ordered phases Multilamellar and hexagonal phases Unilamellar vesicles Micellar systems

#### Continuous - hierarchical - systems

Anisotropy and porosity of activated carbons In situ experiments on a gold-cysteine self-assembling nanocomplex

#### Particulate systems

Size distribution of  $SiO_2$  nanoparticles Proteins – biological macromolecules

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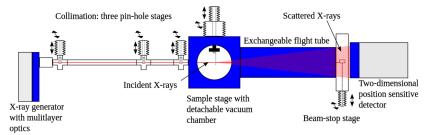
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#### Particulate systems

Size distribution of SiO<sub>2</sub> nanoparticles Proteins – biological macromolecules

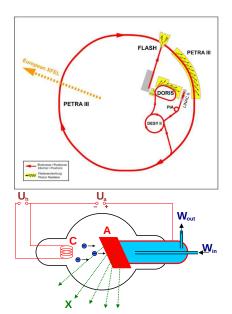
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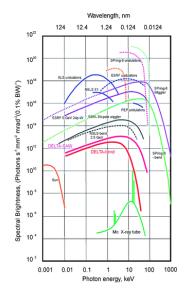
### Pinhole camera



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#### X-ray tube and synchrotron





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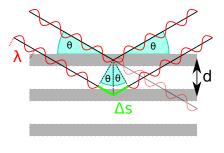
# Bragg's law

- Periodic sample (*d* repeat distance)
- $\theta$  incidence and reflection angle
- Constructive interference at the detector: waves reflected from neighbouring planes meet *in phase*
- $\Delta s = n\lambda$  where  $n \in \mathbb{N}$
- From geometry:  $\Delta s = 2d \sin \theta$

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$$2d\sin\theta = n\lambda$$

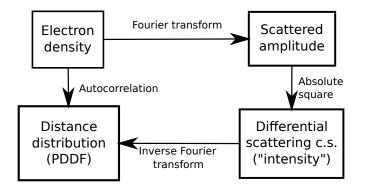
$$\frac{4\pi}{\lambda}\sin\theta = \frac{2\pi}{d}$$

$$q = \frac{2\pi}{d} n$$



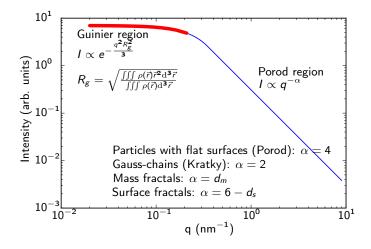
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Connection between structure and scattering



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#### Guinier and Porod limits



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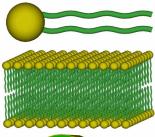
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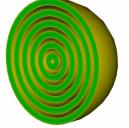
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#### Summary

### Lipid systems, liposomes

- Amphipatic molecules: hydrophilic headgroups, hydrophobic carbon chains
- Self-assemble in aqueous solution
- Cell membranes of living organisms
- Other similar molecules: surfactants, detergents etc.
- Application in research & industry:
  - Model membranes
  - Drug carrier vehicles
  - Nanoreactors
  - **۰**...
- Phase transitions
  - Thermotropic
  - Lyotropic

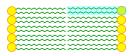


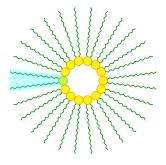


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## Self-assembled structures of phospholipid systems

- The self-assembled structure is determined by:
  - Shape of the lipid molecule
  - Length and flexibility of the carbon chains
  - Electrostatic charge of the headgroups
- Bilayer lipids: approximately cylindrical
- Non-bilayer lipids: conical shape
  - Large headgroup cross-section area: micelle / hexagonal phase
  - Small headgroup cross-section area inverse micelle / inverse hexagonal phase

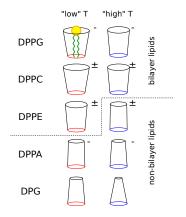






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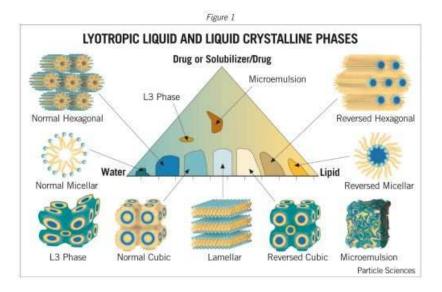
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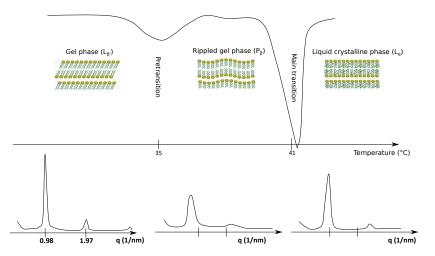
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# Lyotropic phases of lipid/water systems

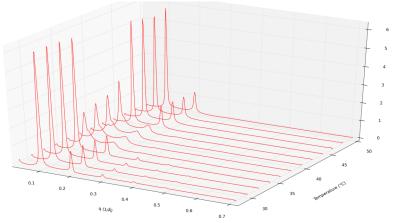


#### Thermotropic phases of DPPC/water mixtures



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### Thermotropic phases of DPPC: SAXS

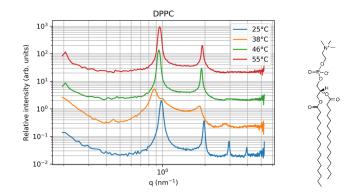


Relative peak positions: 1, 2, 3, 4  $\rightarrow$  lamellar

Intensity (1/cm)

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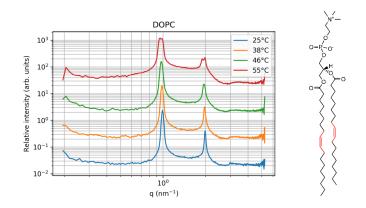
### Thermotropic phases of DPPC: SAXS



Relative peak positions: 1, 2, 3,  $4 \rightarrow$  lamellar

| Temperature     | 25°C      | 38°C         | 46° C      | 55°C       |
|-----------------|-----------|--------------|------------|------------|
| Phase           | $L_{eta}$ | $P_{\beta'}$ | $L_{lpha}$ | $L_{lpha}$ |
| Repeat distance | 6.373 nm  | "7.193 nm*"  | 6.657 nm   | 6.569 nm   |

### DOPC

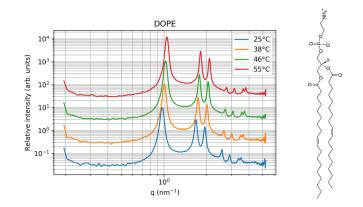


Relative peak positions: 1, 2, (3)

| Temperautre     | 25°C `     | ∕ 38°C     | 46°C       | 55°C       |
|-----------------|------------|------------|------------|------------|
| Phase           | $L_{lpha}$ | $L_{lpha}$ | $L_{lpha}$ | $L_{lpha}$ |
| Repeat distance | 6.323 nm   | 6.370 nm   | 6.440 nm   | 6.642 nm   |
|                 |            |            |            | 6.335 nm   |

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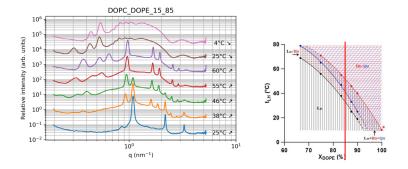
### DOPE: hexagonal phase



Relative peak positions: 1,  $\sqrt{3}$ , 2,  $\sqrt{7}$ , 3,  $\sqrt{12}$ ,  $\sqrt{13}$ 

| Temperature       | 25°C     | 38° C    | 46°C     | 55°C     |
|-------------------|----------|----------|----------|----------|
| Phase             | HII      | HII      | HII      | HII      |
| Lattice parameter | 6.458 nm | 6.244 nm | 6.119 nm | 5.989 nm |

### Coexistence of phases



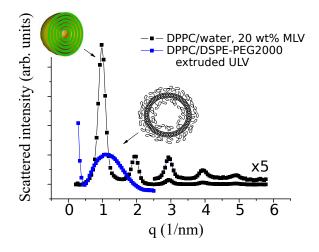
- Room temperature: lamellar phase  $(L_{\alpha})$
- ▶ 38 °C: appearance of the inverse hexagonal phase (H<sub>II</sub>)
- ▶ 46 °C: the cubic phase ( $Q_{II}$ ) appears, three phases coexist
- ▶ 55 °C: the lamellar phase vanishes
- after cooling: the cubic phase remains, the lamellar phase is not recovered: memory effect!

### Sterically stabilized unilamellar vesicles

- Unilamellar vesicle: a single phospholipid bilayer
- Hydration of lipids: multilamellar vesicles are formed *spontaneously*
- "Unilamellarization": ultrasound threatment / extrusion
- Avoiding spontaneous fusion to multilamellar vesicles:
  - Charged lipids
  - Sterical stabilization: e.g. with PEG-conjugated lipids
- ▶ Primary application: drug carrier and targeting agents ⇒ size is critical!

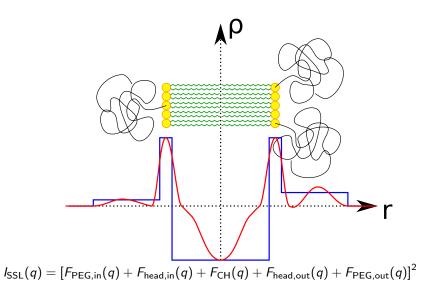
| Ð             | A                | $\mathbb{R}$ |   |
|---------------|------------------|--------------|---|
|               |                  |              |   |
|               |                  |              |   |
|               |                  |              |   |
| $\mathcal{A}$ | $\sum_{i=1}^{i}$ |              |   |
| 6             | Y)               |              | Ŷ |

### Sterically stabilized vesicles



- ► Less electrons in the object ⇒ weaker scattering
- No layer-layer correlation  $\Rightarrow$  no peaks
- ► What we see is the *phospholipid bilayer form factor*

# Scattering of a phospholipid bilayer



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### Scattering of a bilayer

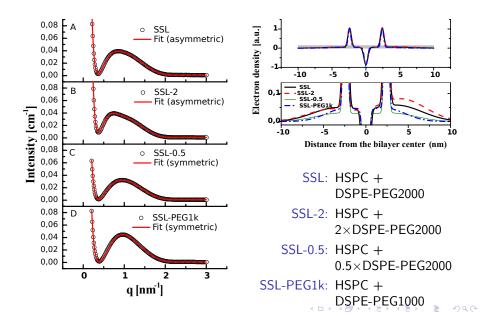
 $I_{\text{SSL}}(q) = [F_{\text{PEG,in}}(q) + F_{\text{head,in}}(q) + F_{\text{CH}}(q) + F_{\text{head,out}}(q) + F_{\text{PEG,out}}(q)]^2$ 

- Every term is a step function or a Gaussian curve
- $\blacktriangleright \rho(q) = \begin{cases} \rho_0 & \text{if } |r r_0| < \sigma \\ 0 & \text{otherwise} \end{cases} \quad \rho(q) = \frac{\rho_0}{\sqrt{2\pi\sigma^2}} e^{-\frac{(r r_0)^2}{2\sigma^2}}$
- Model parameters:

|   | $\rho$           | r                    | $\sigma$            |  |
|---|------------------|----------------------|---------------------|--|
| Inner PEG   | $ ho_{PEG,in}$   | <i>r</i> PEG,in      | $\sigma_{PEG,in}$   |  |
| Inner headgroup   | $ ho_{\sf head}$ | $-r_{head}$          | $\sigma_{\sf head}$ |  |
| Carbon chain  | -1               | 0                    | $\sigma_{tail}$     |  |
| Outer headgroup   | $ ho_{head}$     | $r_{\rm head}$       | $\sigma_{\sf head}$ |  |
| Outer PEG   | $ ho_{PEG,out}$  | r <sub>PEG,out</sub> | $\sigma_{PEG,out}$  |  |
| + global intensity scaling factor $(A)$ + constant background $(C)$ +   |                  |                      |                     |  |
| mean vesicle radius $(R_0)$ + spread of the vesicle radius $(\delta R)$ |                  |                      |                     |  |
|   |                  |                      |                     |  |

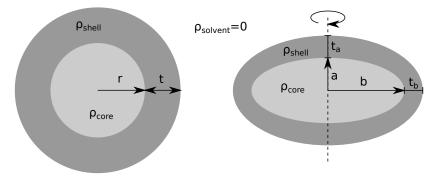
- Asymmetric model (PEGs are different): 14 parameters
- Symmetric model (PEGs are equivalent): 11 parameters

### Sterically stabilized vesicles



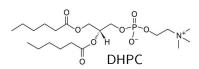
### Micelles

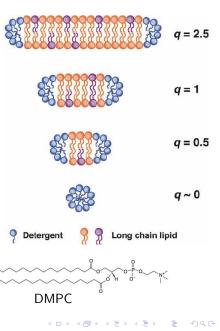
- Self-assembling systems composed of amphipatic molecules
- Conical shape: large hydrophilic head, narrow hydrophobic tail
- Critical micelle concentration (CMC)
- Not only spherical (even when only one component!)



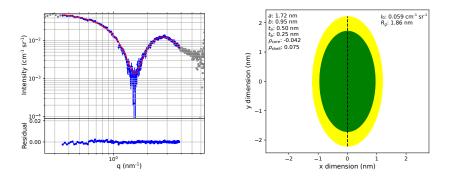
### Bicelles

- Two components: long-chained bilayer lipid and short-chained detergent
- The shape is controlled by:
  - $q = c_{\rm lipid}/c_{\rm detergens}$ 
    - q = 0: detergent micelle
    - $q \to \infty$ : bilayer
- Importance: small carriers for membrane proteins
- Typical example: DHPC-DMPC bicelle
  - DHPC: 1,2-Dihexanoyl-sn-Glycero-3-Phosphocholine
  - DMPC: 1,2-Dimyristoyl-sn-Glycero-3-Phosphocholine



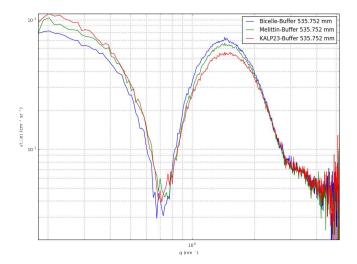


### Scattering of a DHPC micelle



- Scattering: similar to the lipid bilayers
- Guinier region
- Fitting: micelle shape

### Peptide-carrying DHPC-DMPC bicelles



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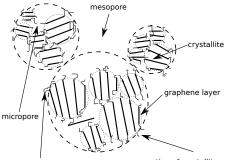
#### Summary

### Activated carbons

#### Activated carbons

- Adsorbent, substrate, structural material
- Hierarchical structure
- Preparation:
  - 1. Pyrolysis: organic  $\rightarrow$  C
  - 2. Activation: pore formation
- Tailorable
  - choice of the precursor
  - parameters of the activation
- Anisotropy: not utilized (but could be...)

#### Model of the hierarchical structure



amorphous carbon aggregation of crystallites Hirsch, Proc. Royal Soc. Lond. A (1954) 226(1165) 143-169

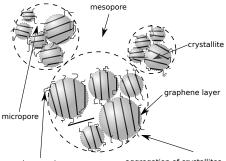
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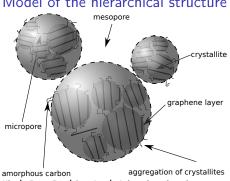
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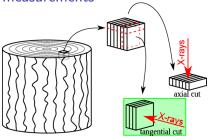
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# Sample preparation

# Sample preparation for SAXS measurements



- ▶ Pyrolysis of 1 cm<sup>3</sup> wooden cubes (700 °C) →  $6 \times 6 \times 6$  mm<sup>3</sup> carbon cubes
- Physical activation:

$$C_{(s)} \xrightarrow{H_2O_{(g)}} C_{(g)}$$

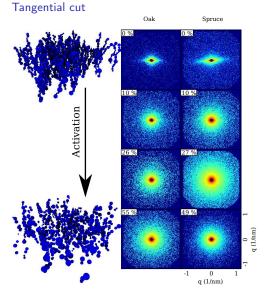
Mass decrease (conversion) with increasing activation time:

|        | Fagus     | Quercus | Picea    |
|--------|-----------|---------|----------|
|        | sylvatica | robur   | abies    |
|        | (beech)   | (oak)   | (spruce) |
| 0 min  | 0%        | 0 %     | 0 %      |
| 15 min | 9 %       | 10 %    | 10 %     |
| 45 min | 26 %      | 26 %    | 27 %     |
| 90 min | 54 %      | 55 %    | 49 %     |

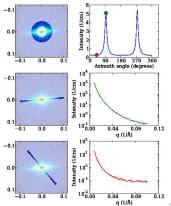
 SAXS measurements: synchrotron beamlines (Hamburg, Berlin)

For details, see: Wacha, Varga, Vainio, Hoell, Bóta (2011) Carbon 49(12) 3958-3971.

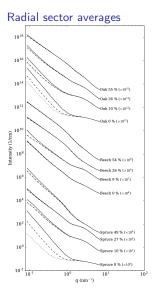
# SAXS on activated carbons



- Horizontal scattering pattern: vertical fibrils
- Decrease in anisotropy: breaking of the fibrils, pore formation
- Characterization of anisotropy: azimuthal scattering curves, sector averaging



# Extent of the anisotropy in real space

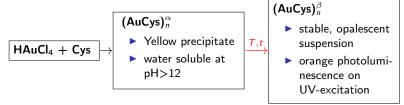


Radial scattering curves from the scattering patterns

- Averaging over the full  $2\pi$  (- -)
- Narrowed to the region of the most intensive azimuthal peak (—)
- Perpendicular to the previous direction  $(\cdots)$
- Anisotropy does not appear at small sizes  $(q > 2 \text{ nm}^{-1} \rightarrow d < 1.5 \text{ nm})$
- Anisotropy decreases with activation
- Power-law functions (→ fractal dimension) and Guinier regions (→ radius of gyration)
- Two Guinier regions
  - Small conversion (short activation time): micropores
  - Large conversion (long activation time): mesopores
- Mass fractal  $\rightarrow$  surface fractal transition
  - Spruce: surface fractal appears after 49 % burn-off: microcracks
  - Beech: no surface fractal: inherently porous?

### Photoluminescent gold-cystein nanocomplexes

- Protein-stabilized supramolecular gold clusters: photoluminescence
- Au-Cys nanocomplex: a simple model for uncovering the stabilizing mechanism

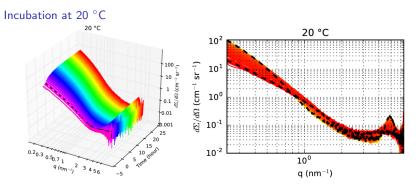


► The speed of transition strongly depends on the temperature of incubation, ranging from a few hours to a day ⇒ time-resolved SAXS on CREDO

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Söptei et. al. 2015 Coll.Surf.A 470, 8-14.

## TRSAXS on the Au-Cys nanocomplex



Curvature at small  $q \rightarrow$  Guinier

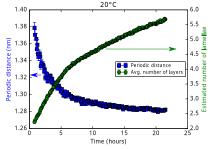
- Objects with well-defined sizes
- Moves left  $\rightarrow$  increase in size
- ► Increasing intensity → their number increases
- Starts with I ∝ q<sup>-2</sup> → thin lamellae (generalized Guinier)

Peak at the high-q limit

- Periodic structure
- ► Increasing intensity → more perfect periodicity

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# Automated model fitting

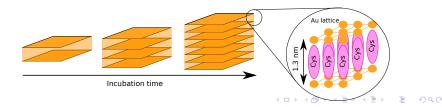


#### Number of layers and periodicity

 Guinier approximation for extended lamellae:

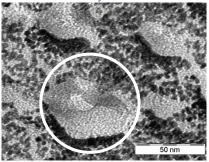
 $I_{\text{thickness}} \approx G \cdot q^{-2} e^{-q^2 R_T^2} \rightarrow$ thickness of the homogeneous lamella:  $T = \sqrt{12} R_T$ 

- ▶ Final periodic distance: 1.29 nm
   ▶ Fine structure of the lamellae: Au layers above each other with ≈ 1.3 nm distance, the Cys molecules acting as spacers
- Well-correlated with the increase of photoluminescence intensity: 0.9208



# Automated model fitting

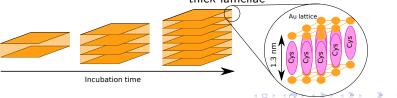
Lamellae as seen by TEM



 Guinier approximation for extended lamellae:

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- Final periodic distance: 1.29 nm
   Fine structure of the lamellae: Au layers above each other with ≈ 1.3 nm distance, the Cys molecules acting as spacers
- Well-correlated with the increase of photoluminescence intensity: 0.9208
- ► FF-TEM measurements: a few nm thick lamellae



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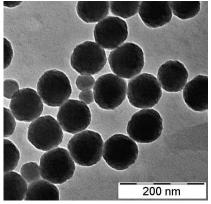
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### Summary

## Size distribution of SiO $_2$ nanoparticles

Institute for Reference Materials and Measurements, Joint Research Centre of the European Commission: introducing a new SiO<sub>2</sub> particle size standard. Certification of the new material with several SAXS instruments

# ERM FD-101b: candidate reference material (CRM)



- Methods of size determination:
  - 1. Guinier fit:  $I(q \ll 1/R) \approx I_0 e^{-\frac{q^2 R^2}{5}}$ 2. Fitting of the sphere form factor:  $I(q) = \Phi_{\text{sphere}}(q, R) \equiv V_R^2 \left[\frac{3}{(qR)^3} (\sin(qR) - qR\cos(qR))\right]^2$

3. Fitting of a sphere distribution:  

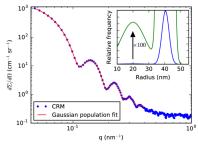
$$I(q) = \int_{0}^{\infty} p(R) \Phi_{\text{sphere}}(q, R) dR$$
4. Monte Carlo method:  $R_i$  population

with  $w_i$  statistical weights where  $|I(q) - \sum_i w_i \Phi_{sphere}(q, R_i)|$  is minimized

## Size distribution of $SiO_2$ nanoparticles

Institute for Reference Materials and Measurements, Joint Research Centre of the European Commission: introducing a new SiO<sub>2</sub> particle size standard. Certification of the new material with several SAXS instruments





- Methods of size determination:
  - 1. Guinier fit:  $I(q \ll 1/R) \approx l_0 e^{-\frac{q^2 R^2}{5}}$ 2. Fitting of the sphere form factor:  $I(q) = \Phi_{\text{sphere}}(q, R) \equiv V_R^2 \left[\frac{3}{(qR)^3} (\sin(qR) - qR\cos(qR))\right]^2$ 3. Fitting of a sphere distribution:

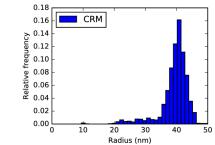
Fitting of a sphere distribution:  

$$I(q) = \int_{0}^{\infty} p(R) \Phi_{\text{sphere}}(q, R) dR$$

4. Monte Carlo method:  $R_i$  population with  $w_i$  statistical weights where  $|I(q) - \sum_i w_i \Phi_{sphere}(q, R_i)|$  is minimized

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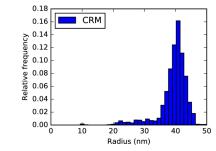
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A favourable side-effect: CREDO has been certified by IRMM for nanoparticle size distribution determination

# Biological Small-Angle X-ray Scattering

## BioSAXS

- Biorelevant macromolecules
- Mainly size- and shape determination assuming particles of homogeneous electron density
- Key parameters:  $R_g$ ,  $I_0 \equiv \lim_{q \to 0} I(q)$
- Information to be obtained:
  - Size, (low resolution) shape, volume and molecular mass of the protein
  - Flexibility/folding state (folded/disordered)
  - Validation of crystal structures
  - Aligning the relative positions of known domains

## Drawbacks / caveats

- ► Low scattering contrast ⇒ bad signal/noise ratio
- Dilute sample (otherwise Guinier approximation breaks down)
- Purified sample (esp. contaminating large molecules)
- Monodisperse sample (avoid oligomerization, aggregation)
- Featureless scattering curve: danger of "overfitting"
- Uncertainties of background subtraction (solvent scattering)
- ► Phase problem ⇒ the uniqueness of the determined shape

## The BioSAXS method

- Well-established and validated algorithms and methods available
- Basic assumptions: the protein solution is a monodisperse population of independent, homogeneous nanoparticles independent: no interparticle interference, Guinier approximation holds
   monodisperse: no oligomerization, no aggregation homogeneous: simple shape fitting; SAXS is blind on the atomic length-scale!



## Interpretation of BioSAXS measurements

- Guinier approximation:  $I(q \ll R_g) \propto l_0 e^{-\frac{q^2 R_g^2}{3}}$ ;  $l_0 = (\Delta \rho)^2 V^2$ .
- Porod invariant:  $Q \equiv \frac{1}{2\pi^2} \int_{0}^{\infty} q^2 I(q) dq = 2\pi^2 (\Delta \rho)^2 V$
- Porod volume:  $V_{\text{Porod}} = 2\pi^2 I_0 / Q$
- First steps:
  - 1. Subtraction of the solvent background (corrected by the volume fraction of the protein)
  - 2. Guinier fit  $\rightarrow I_0, R_g$
  - 3. Porod invariant  $\rightarrow V_{\mathsf{Porod}}$
  - 4. Inverse Fourier:  $I(q) \rightarrow p(r)$  pair distance distribution function (PDDF)
  - 5.  $I_0$ ,  $R_g$  can be obtained from p(r):

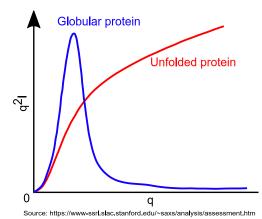
$$I_0 = \int_0^\infty p(r) \mathrm{d}r; \qquad R_g^2 = \frac{\int_0^\infty p(r) r^2 \mathrm{d}r}{2 \int_0^\infty p(r) \mathrm{d}r}$$

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- 6. Compare the  $I_0$  and  $R_g$  obtained from the two methods
- 7. Further interpretation...
- ATSAS: software suite for BioSAXS data processing and interpretation (EMBL Hamburg, Research Group of Dmitri Svergun)

## The Kratky plot

- ► High-*q* part of the scattering of a polimer chain following Gaussian statistics:  $I(q \rightarrow \infty) \propto \frac{2}{q^2 R_z^2}$
- Kratky plot:  $q^2I$  q. Behaviour in the  $q \rightarrow \infty$  limit:
  - Folded proteins  $(I \propto q^{-4})$ : tends to 0
  - Disordered proteins  $(I \propto q^{-2})$ : constant or divergent



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## Protein shape fitting from small-angle scattering

## Fitting of geometrical shapes to scattering curves or PDDFs

- BODIES program (part of ATSAS)
- Ball, hollow sphere, ellipsoid, dumbbell etc.
- Very few parameters

## Dummy atom model (DAM)

- Constructing the shape from tightly packed (fcc or hcp lattice) spherical building blocks
- Monte Carlo algorithm
  - 1. Random configuration
  - 2. Small, random modification of the configuration (add/remove a unit)
  - 3. Calculate scattering
  - 4. Compare the measured and calculated scattering
    - Better fit: keep the change
    - Worse fit: drop the change (or keep it with a low probability)
  - 5. Repeat from step #2 until needed
- Many parameters: possible ambiguity of the results

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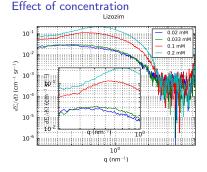
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#### Crystal structure

Well-known protein ("veterinary horse")

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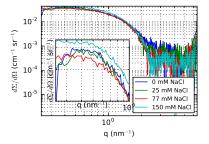
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- How to get rid of it?
  - Dilution

## Effect of ionic strength

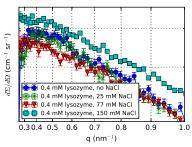


- Well-known protein ("veterinary horse")
- Correlation peak
  - Caused by el.stat. repulsion
  - Radius of gyration cannot be determined
  - How to get rid of it?
    - Dilution
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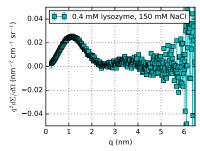
## Guinier plot



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► Guinier plot (log *I* vs. *q*<sup>2</sup>): assessing the *I* ∝ exp(-*q*<sup>2</sup>*R*<sup>2</sup><sub>g</sub>/3) shape

## Kratky plot

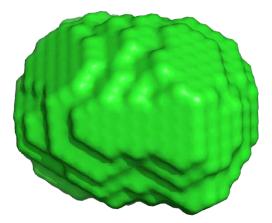


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- Kratky plot  $(q^2 I \text{ vs. } q)$ : folded protein

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## The shape of lysozyme

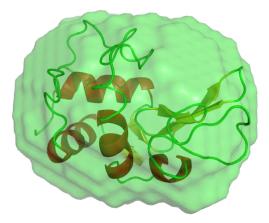
"Dummy atom model" - coarse-grained description



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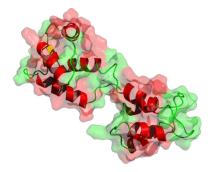
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Good agreement with the crystal structure!

## Calmodulin

- ► Highly abundant plasma protein of eukaryotic cells (≈ 1 %)
- Key element of Ca<sup>2+</sup>-induced signal pathways

Apo (Ca<sup>2+</sup>-free) conformation (MX)



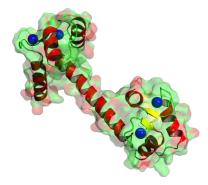
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#### Envelope: Van der Waals surface

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- ► Highly abundant plasma protein of eukaryotic cells (≈ 1 %)
- Key element of Ca<sup>2+</sup>-induced signal pathways
- Changes shape on Ca<sup>2+</sup> binding
  - The "EF-hand" motifs open in both end-domains: hydrophobic pockets open up
  - End domains are displaced
  - Secondary structure of the linker part: loop → helix (known crystallization artefact!)

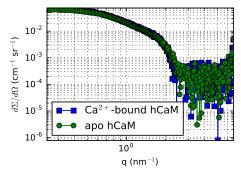
# Ca<sup>2+</sup>-bound conformation (MX)



Envelope: Van der Waals surface

# Calmodulin – SAXS results

## Scattering curves



Very similar scattering curves

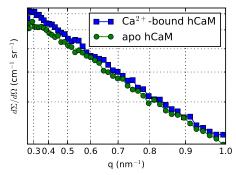
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Scattering curves: dumbbell shape

# Calmodulin – SAXS results

## Guinier plot



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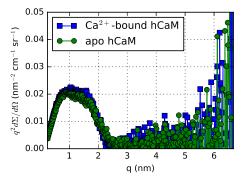
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- Scattering curves: dumbbell shape
- Similar radii of gyration

# Calmodulin – SAXS results

## Kratky plot



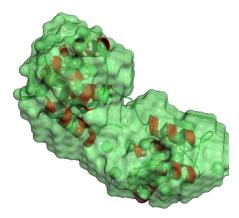
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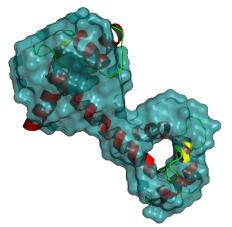
## DAM: apo conformation



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- Scattering curves: dumbbell shape
- Similar radii of gyration
- Partially disordered (linker part?)
- Dummy atom model:
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  - Apo conformation more "loose"

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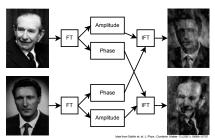
Calmodulin – SAXS results DAM: Ca<sup>2+</sup>-bound conformation



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- Partially disordered (linker part?)
- Dummy atom model:
  - Dumbbell shape
  - Apo conformation more "loose"
  - Ca<sup>2+</sup> binding makes the structure more rigid
  - Differences from the crystal structure: crystallization artefacts?

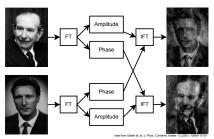
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- Phase problem!
- Methods to improve reliability
  - Several candidate shapes from multiple runs of DAMMIF
  - Screening the candidates with DAMSEL
  - Average the remaining shapes with DAMAVER
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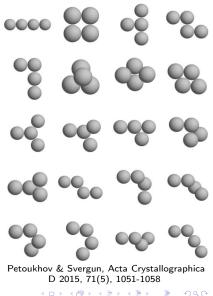
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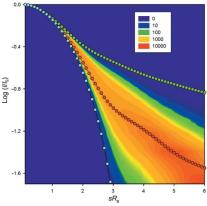


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- Phase problem!
- Methods to improve reliability
  - Several candidate shapes from multiple runs of DAMMIF
  - Screening the candidates with DAMSEL
  - Average the remaining shapes with DAMAVER
  - Refine the average shape with DAMMIN
- Quantification of the ambiguity (AMBIMETER)
  - A library has been made from all possible shapes
  - Dimensionless scattering curves for the library elements: I(q)/I<sub>0</sub> vs. qR<sub>g</sub>

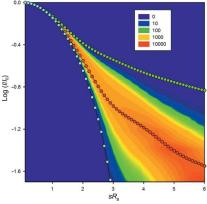


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  - Lysozyme: 1; apo calmodulin: 422; Ca<sup>2+</sup>-bound calmodulin: 417



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## Outline

## Recapitulation

## Self-assembled systems

Ordered phases Multilamellar and hexagonal phases Unilamellar vesicles Micellar systems

### Continuous – hierarchical – systems

Anisotropy and porosity of activated carbons In situ experiments on a gold-cysteine self-assembling nanocomplex

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#### Particulate systems

Size distribution of SiO<sub>2</sub> nanoparticles Proteins – biological macromolecules

## Summary

## Literature and software

## Software

- SASFit: model fitting
- ► ATSAS: BioSAXS data handling, *R<sub>G</sub>*, PDDF calculation, dummy atom fitting etc.
- SANSView: plotting, model fitting

## Literature

- Boualem Hammouda: Probing Nanoscale Structures: The SANS Toolbox (http://www.ncnr.nist.gov/staff/hammouda/the\_ SANS\_toolbox.pdf)
- J. Kohlbrecher, I. Breßler: SASFit manual (http://kur.web.psi.ch/sans1/SANSSoft/sasfit.html)
- L. A. Feigin és D. I. Svergun: Structure Analysis by Small-Angle X-Ray and Neutron Scattering (http://www.embl-hamburg.de/ biosaxs/reprints/feigin\_svergun\_1987.pdf)

## Summary

Interpretation of SAXS results

- Multilamellar vesicles and ordered lipid systems: determination of the periodic repeat distance
- Sterically stabilized vesicles: the radial electron density distribution of the phospholipid bilayer
- Micelles and bicelles: shape, core-shell model parameters
- Activated carbons: anisotropy, fractal properties
- Gold-cysteine nanocomplex: the time evolution of the photoluminescent nanostructure
- ▶ SiO<sub>2</sub> nanoparticles (repeated): size, size distribution
- BioSAXS: determination of the size, shape and flexibilitx of proteins in solution

## Acknowledements

The experimental results presented in this lecture could not have been made without the contributions (sample preparations, ideas etc.) of the following people:

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- Zoltán Varga
- Andrea Jónás
- Andrea Bodor
- Erika Dudás
- Tünde Juhász
- Balázs Söptei

## Thank you for your attention!

